

Asymmetric reduction of enones with *Synechococcus* sp. PCC 7942

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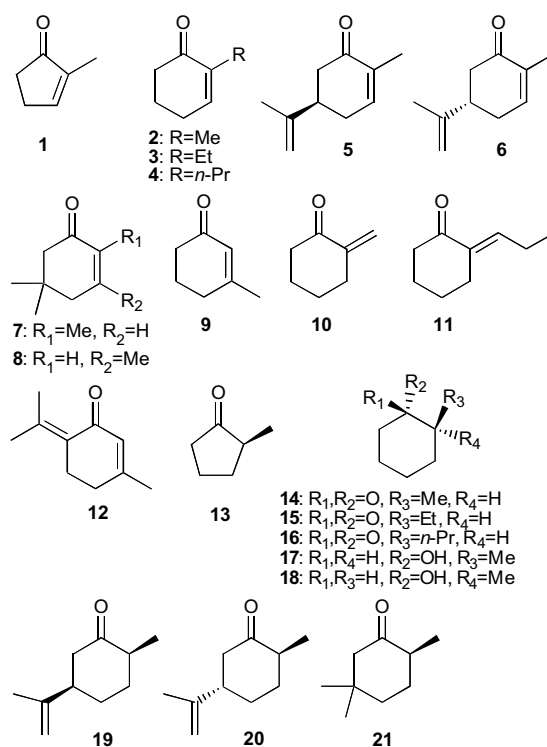
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Abstract—*Synechococcus* sp. PCC 7942, a cyanobacterium, reduced both the endocyclic C–C double bond of *s-trans* enones and the exocyclic C–C double bond of *s-cis* enones with high enantioselectivity to afford the corresponding (*S*)-ketones under illumination. © 2004 Elsevier Ltd. All rights reserved.

Optically active α -substituted ketones are versatile chiral building blocks for asymmetric synthesis.¹ Asymmetric reduction of enones by living whole cells is very attractive and useful for the practical preparation of α -chiral ketones because of high enantioselectivity, no cofactor requirement and ease of scaling up. Microorganisms and plants capable of reducing *s-trans* enones to (*R*)-ketones have been reported so far.^{2–4} However, a cell-mediated process in which *s-trans* enones are reduced to (*S*)-ketones is still unavailable. On the other hand, yeast-catalyzed reduction of *s-cis* enones afforded (*S*)-ketones.⁴ Over the course of developing a new cell-mediated reduction for asymmetric induction, we investigated the asymmetric reduction of enones by *Synechococcus* sp. PCC 7942.

First, *s-trans* enones **1–9** (10 mg each) with an endocyclic C–C double bond were administered to 50 mL of a suspension of *Synechococcus* sp. PCC 7942 cells (2 g)^{5,6} in 50 mM Na-phosphate buffer (pH 7.0) and incubated at 25 °C for 1 or 3 days under illumination.⁷ The yields of the products were determined by GLC analyses. Extraction from the cell broth with ether followed by purification using column chromatography on silica gel with pentane/ethyl acetate (95:5, v/v) gave the products. It was found that the C–C double bonds of **1–3** were reduced to give the corresponding (*S*)-ketones, as shown in Table 1.^{8–11} The enantiomeric purities of the resulting ketones were determined based on the peak area of the corresponding enantiomers in the GLC analyses on CP cyclodextrin β 236M-19.¹² Enone **1** was the best sub-



strate, allowing us to achieve the highest enantiomeric excess (98% ee) and yield (>99%). In the case of **2**, the reduction of the C–C double bond of **2** was accompanied by the formation of minor saturated (*S*)-alcohols **17** (>99% ee in 7% yield) and **18** (>99% ee in 2% yield).^{13–15} No reduction occurred in the case of **4**, which had an

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Table 1. Reduction of enones by *Synechococcus* sp. PCC 7942

Substrates	Products	Reaction time (day)	Conversion (%) ^a	Ee (%)	Configuration ^b
1	13	1	>99	98	<i>S</i>
2	14	1	86	85	<i>S</i>
3	15	3	17	83	<i>S</i>
4	—	3	0	—	—
5	19	3	>99	80 ^c	<i>S</i>
6	20	3	>99	81 ^c	<i>S</i>
7	21	3	15	86	<i>S</i>
8	—	3	0	—	—
9	—	3	0	—	—
10	14	1	82	71	<i>S</i>
11	16	1	7	72	<i>S</i>
12	—	3	0	—	—

^a Percentage of the products in the reaction mixture on the basis of GLC analyses.^b Preferred configuration at the α -position to the carbonyl group of the products.^c Diastereomeric excess.

n-propyl group as the α -substituent. After three days incubation, **5**–**7** were reduced to the corresponding (*S*)-ketones **19**–**21**. *Synechococcus* sp. PCC 7942 cells were not able to reduce β -substituted substrates **8** and **9**. The results obtained here reveal that *Synechococcus* sp. PCC 7942 cells have (i) the ability of catalyzing enantioface differentiating reduction of *s-trans* enones to afford (*S*)-ketones and (ii) similar substrate specificity to microorganisms, which reduce *s-trans* enones if the substituent at the β -position to the carbonyl group is hydrogen and if the α -substituent is not too bulky.^{2,4}

Next, *s-cis* enones **10** and **11**^{16,17} with an exocyclic C=C double bond were subjected to the same reduction system. **10** was smoothly reduced to give (*S*)-ketone **14** in 82% yield, and the hydrogenation at the α -position showed relatively low enantioselectivity (71% ee). Saturated alcohols **17** (>99% ee) and **18** (>99% ee) were formed as minor products in 7% yield (4:1). The reduction of **11** gave (*S*)-ketone **16** with 72% ee in 7% yield. On the other hand, substrate **12**, which had both endocyclic and exocyclic C=C double bonds was not reduced by the cells probably due to the existence of the β -methyl group. These results demonstrate that *Synechococcus* sp. PCC 7942 cells have the same enantioselectivity in the reduction of *s-cis* enones as yeast.⁴

Thus, the asymmetric reductions of *s-trans* and *s-cis* enones have been accomplished and optically active α -substituted (*S*)-ketones have been prepared by using *Synechococcus* sp. PCC 7942 as biocatalyst. It is worth noting that this new biocatalyst has opposite enantioselectivity in the reduction of *s-trans* enones to other microorganisms^{2,4} and plants³ and that each enantiomer of the α -substituted ketones can be synthesized by selective use of the whole cells. Recently, two enone reductases have been isolated from *Nicotiana tabacum*; *s-trans* enone reductase, which was responsible for the reduction of the endocyclic C=C double bond and *s-cis* enone reductase, which was capable of reducing the exocyclic C=C double bond.¹⁸ In *Synechococcus* sp. PCC 7942 *s-trans* enone reductases with an opposite enantioselectivity to those from yeast and *N. tabacum* might exist. Further investigations using the enzyme prepara-

tion from *Synechococcus* sp. PCC 7942 are currently in progress.

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- reduced to **13** with 70% ee in 42% yield after one day's incubation in the dark.
8. Product **13**: $[\alpha]_{\text{D}}^{25} = +114.9$ (c 0.52, CHCl_3) {lit.⁹ $[\alpha]_{\text{D}}^{25} = -110.5$ for (*R*)-enantiomer}; **14** converted from **2**: CD $[\theta]_{288} = +887$ (c 0.75, MeOH) {lit.¹⁰ $[\theta]_{288} = -987$ for (*R*)-enantiomer}; **14** from **10**: CD $[\theta]_{288} = +701$ (c 0.68, MeOH); **15**: CD $[\theta]_{288} = +1914$ (c 0.32, MeOH) {lit.¹¹ $[\theta]_{288} = +2200$ }; **16**: CD $[\theta]_{288} = +1860$ (c 0.15, MeOH) {lit.¹¹ $[\theta]_{288} = +2480$ }; **21**: CD $[\theta]_{288} = +995$ (c 0.14, MeOH).
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12. Conditions for capillary GLC analysis: column, CP cyclodextrin β 236M-19 (0.25 mm \times 25 m); injection, 180 °C; detector, 180 °C; oven, 100 °C; carrier gas, N_2 (50 mL min⁻¹). Retention times for the products in the GLC were as follows: (*S*)- and (*R*)-**13**, 10.5 and 11.4 min; (*S*)- and (*R*)-**14**, 11.8 and 12.8 min; (*S*)- and (*R*)-**15**, 12.7 and 12.9 min; (*S*)- and (*R*)-**16**, 27.7 and 27.9 min; (*S*)- and (*R*)-**21**, 50.1 and 51.9 min.
13. Product **17**: $[\alpha]_{\text{D}}^{25} = +51.2$ (c 0.4, MeOH) {lit.¹⁴ $[\alpha]_{\text{D}}^{20} = +42.9$ }; **18**: $[\alpha]_{\text{D}}^{25} = +25.7$ (c 0.2, MeOH) {lit.¹⁴ $[\alpha]_{\text{D}}^{20} = +24.3$ }. The enantiomeric purities of **17** and **18** were determined by ¹H NMR analyses of the corresponding MTPA esters as described previously.¹⁵ In the cases of the other substrates, saturated alcohols were not obtained during the incubation time examined.
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